

Atmospheric Pressure Chemical Ionisation as an alternative ion source for the analysis of problematic pesticides by GC-QQQ Mass Spectrometry.

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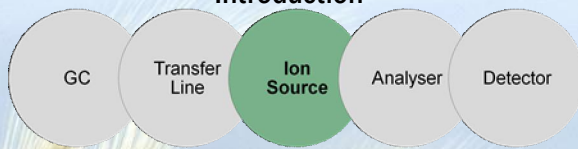


Summary

- ~ 1. Introduction
- ~ 2. GC Method
- ~ 3. Ionisation processes
- ~ 4. Difficult pesticides
- ~ 5. Method Development . Fruit & Vegetables
- ~ 6. Method Validation
- ~ 7. APCI v EI - sample comparison
- ~ 8. Conclusion and future developments

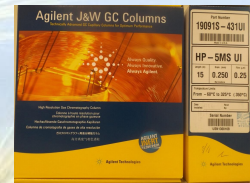


Introduction



GC Method

- ~ Agilent J&W HP5 MS UI column
- ~ 15m with retention gap
- ~ 0.25mm (narrowbore) x 250mm
- ~ Pulsed splitless injection
- ~ 2µl injection volume
- ~ Constant flow



GC Method

Inlet program

GC Temperature program

Inlet program

Settings: Inlet Type: Methods, Mode: Pulsed Splitless, Flow Rate: 15.0 mL/min, Temperature: 30 °C, On Time: 3.00 min, Row: 15.0 min.

GC Temperature program

Run	Temp	Time	Temp	Time	Temp	Time
1	20.00	1.00	1.00	20.00	1.00	2.00
2	6.00	2.00	0.00	11.93	6.00	2.00
3	16.00	2.00	0.00	11.93	16.00	2.00
4	6.00	2.00	0.00	11.93	6.00	2.00
5	6.00	2.00	0.00	11.93	6.00	2.00
6	6.00	2.00	0.00	11.93	6.00	2.00
7	6.00	2.00	0.00	11.93	6.00	2.00
8	6.00	2.00	0.00	11.93	6.00	2.00
9	6.00	2.00	0.00	11.93	6.00	2.00
10	6.00	2.00	0.00	11.93	6.00	2.00



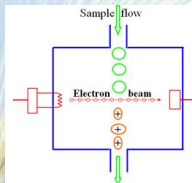
Introduction

Primary ion production

1. Electron impact (EI) ionisation
2. Chemical Ionisation . Positive and Negative (PCI & NCI)
3. Atmospheric Pressure Chemical Ionisation (APCI)

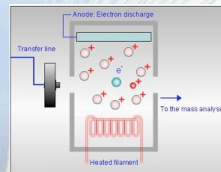


Electron Impact Ionisation



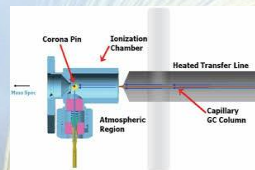
1. The sample is introduced into the ion source from the GC . under vacuum
2. An electron beam (70 eV) passes through the sample causing ionisation / fragmentation
3. The ions are focussed into a beam and pass into the analyser.
4. In the case of QQQ secondary fragmentation occurs in the collision cell.

Chemical Ionisation



1. The sample is introduced into the ion source from the GC . under vacuum
2. The sample is mixed with a reagent gas such as NH_4
3. An electron beam (70 eV) passes through the reagent gas causing ionisation of the gas
4. Collisions between the ionised reagent gas and the sample causes ionisation of the sample
3. The ions are focussed into a beam and pass into the analyser.
4. In the case of QQQ secondary fragmentation occurs in the collision cell.

Atmospheric Pressure Chemical Ionisation



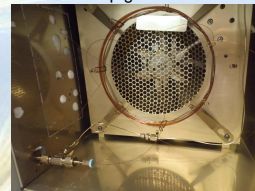
1. The effluent from the GC is mixed with a make-up gas . high purity N_2
2. The gaseous mixture passes over a charged corona pin which induces ionisation in the make-up gas producing an N_2^+ plasma
3. Collisions between the charged N_2^+ molecules and the sample induce ionisation in the sample

Atmospheric Pressure Chemical Ionisation

APCI Source



CI Make up gas connections



Atmospheric Pressure Chemical Ionisation

Advantages:

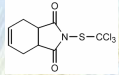
1. Softer ionisation
2. A higher probability of more unique ions
3. Can switch between GC and HPLC in minutes



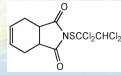
Comparison of ionisation mechanisms

	EI	CI	APCI
Ionisation	Hard	Soft	Soft
Molecular ion formation	Poor	OK	Good
Charge	+	+ / -	+

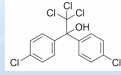
The Good, The Bad & The Ugly



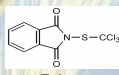
Captan



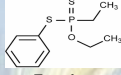
Captafol



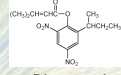
Dicofol



Folpet



Fonofos



Binapacryl

The mix from Hell !!

Why are these pesticides difficult ?

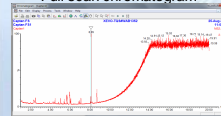
- ~ 1. Liner effects:
 - ~ Active liners can adsorb some difficult pesticides
- ~ 2. Fragmentation/degradation:
 - ~ Primary ionisation results in ions which are not necessarily unique
- ~ 3. Matrix:
 - ~ Interference

Method Development

- ~ 1. MSScan method to determine the Retention Time & the parent ion
- ~ 2. Daughter ion scan to determine possible product ions
- ~ 3. Optimisation of cone voltage and collision energy
- ~ 4. Build MS Method & check linearity
- ~ 5. Validation

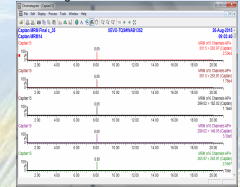
Captan

Full scan chromatogram

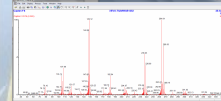


RT at 8 mins

Results of Daughter ion scanning showing the transitions

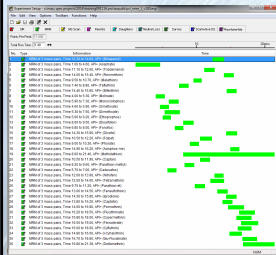


MRM Method TIC

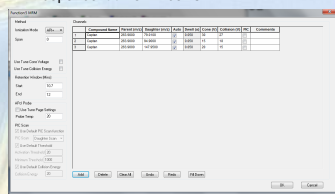


Select parent ion at 264

MS Method



Optimised transitions in compound dependent time windows



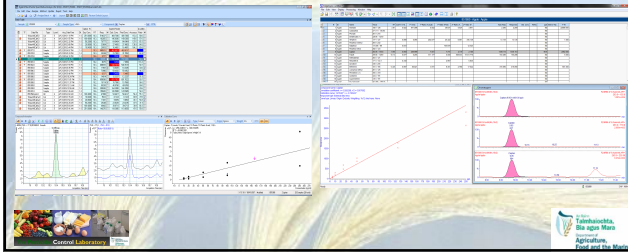
Comparison of Transitions

Pesticide	Mass	APCI	EI
Captan	300.6	263.9 → 79.01	79 → 77
		263.9 → 94.9	79 → 51
		263.9 → 147.95	
Captafol	349.1	311.78 → 79.0	150.9 → 78.9
		311.78 → 116.86	150.9 → 51.0
		311.78 → 160.81	150.9 → 150.9
Dicofol	370.5	252.92 → 112.98	139 → 111
		252.92 → 138.96	139 → 75
		252.92 → 140.96	
Folpet	296.6	259.81 → 94.91	147 → 103.1
		259.81 → 102.01	147 → 76
		259.81 → 129.99	
Fonofos	246.3	246.342 → 80.92	246 → 109.1
		246.344 → 125.99	246 → 81.1
		246.346 → 108.97	
Binapacryl	290.8	222.02 → 146.03	82.9 → 55.1
		222.02 → 158.02	82.9 → 39
		222.02 → 175.04	

In general APCI transitions are more unique than those in EI

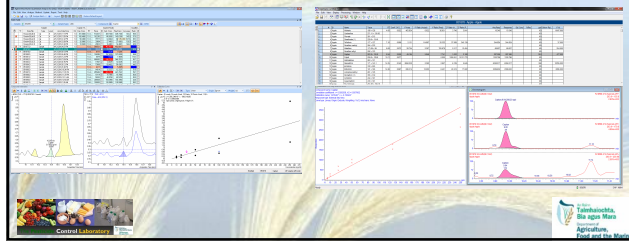
Captan - Apple comparison

Reasonable EI data confirmed by APCI



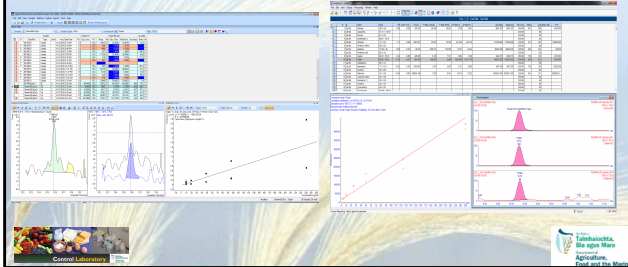
Captan - Apple comparison 2

Very poor EI data but APCI data excellent . possible false negative avoided !!



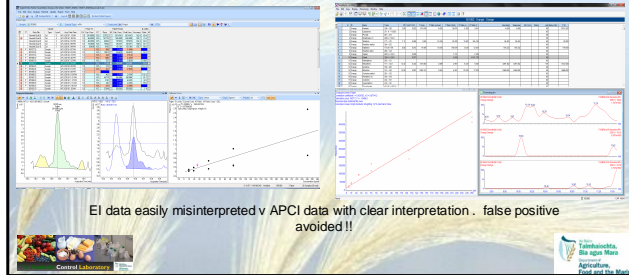
Folpet - Standard Comparison

Poor EI data v excellent APCI data



Folpet - Orange Comparison

EI data easily misinterpreted v APCI data with clear interpretation . false positive avoided !!



Future work

- Expand the method to:
 - Cereals
 - Milk and eggs
 - Honey
 - Infant formula
 - Fats
- Expand scope of pesticides and metabolites
- Accreditation

Conclusion

- Have we tamed the mix from Hell ?
 - Probably not but they are much calmer and the data is much more reliable with this method
- This work is in its infancy . work in progress
- The number of false positive and false negatives is reduced greatly
- This technique has potential
- Could also work for difficult LC compounds

Thank you for your attention

Any Questions?

